# Preparation and Transport Properties of New Ternary Tellurides $A_x Nb_6 Te_8$ (A = Tl, Ca, Sr, Ba, La, and Nd)

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New ternary niobium tellurides  $A_x Nb_6 Te_8$  (A = Ca, Sr, Ba, La, and Nd; x = 0.22-0.32) have been prepared by molten salt ion exchange reactions of  $Tl_x Nb_6 Te_8$  with rare earth/alkaline earth iodides  $AI_3/AI_2$  at relatively low temperatures, 535– 795°C. Most of the niobium tellurides show metallic behavior followed by superconductivity at low temperatures, 4–6 K. Polycrystalline  $La_{0.32}Nb_6Te_8$  and  $Nd_{0.24}Nb_6Te_8$  show a resistivity anomaly at 46 and 43 K, respectively, while polycrystalline  $Tl_{0.91}Nb_6Te_8$  shows a broad resistivity anomaly at 160–220 K. The resistivity anomaly observed in  $Tl_{0.91}Nb_6Te_8$  appears to be accompanied by a minor structural distortion evidenced in differential scanning calorimetry. The resistivity anomalies observed in  $La_{0.32}Nb_6Te_8$  and  $Nd_{0.24}Nb_6Te_8$  are similar to those observed at ~40 K for  $Nb_3Te_4$  and may be related to the onset of a charge density wave type instability. (2) 1996 Academic Press, Inc.

#### INTRODUCTION

Recently, considerable effort has been directed toward the understanding of physicochemical properties of transition metal chalcogenides (1–6). Among many transition metal chalcogenides, both the binary and ternary niobium chalcogenides Nb<sub>3</sub> $Q_4$  and  $A_x$ Nb<sub>6</sub> $Q_8$  (A = Alkali, Tl, In, etc; Q = S, Se, and Te) have attracted much interest, due to a number of interesting properties such as superconductivity, charge density wave (CDW) instabilities, and ion intercalation (7–11).

Nb<sub>3</sub> $Q_4$  crystallizes in a hexagonal unit cell (space group  $P6_3/m$  or  $P6_3$ ) (7, 12). The crystal structure of Nb<sub>3</sub> $Q_4$  is built up from face sharing of three Nb $Q_6$  octahedra and by edge sharing of the resulting Nb<sub>3</sub> $Q_{11}$  groups (Fig. 1) (13). Each Nb atom in the Nb $Q_6$  octahedra is displaced (by ~0.3 Å for Q = S) toward one of the faces of the octahedron, which results in the formation of infinite zig-zag metal chains along the *c* axis (12). The shortest Nb–Nb intrachain distances are 2.88(1), 2.80(1), and 2.97(1) Å for Q = S, Se, and Te, respectively, which are comparable to

those in metallic Nb( $d_{\text{Nb-Nb}} = 2.86$  Å). On the other hand, the interchain Nb–Nb distances between face-shared octahedra are considerably longer, 3.37(1), 3.47(1), and 3.65(1) Å, respectively, suggesting a quasi-one-dimensional nature of the electronic structure. Another important structural feature of the Nb<sub>3</sub>Te<sub>4</sub>-type compounds is the hexagonal tunnels running along the *c* axis. The empty tunnels in Nb<sub>3</sub>Q<sub>4</sub> are large enough to accommodate ions in the voids.

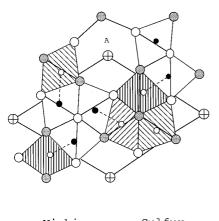
Nb<sub>3</sub>S<sub>4</sub>, Nb<sub>3</sub>Se<sub>4</sub>, and Nb<sub>3</sub>Te<sub>4</sub> exhibit metallic conductivity ( $\rho_{\text{RT}} = 10^{-3}-10^{-4} \ \Omega \cdot \text{cm}$ ) as well as a superconducting transition at 4.0, 2.0, and 1.8 K, respectively (8). Ishihara and Nakada observed two types of phase transitions in a Nb<sub>3</sub>Te<sub>4</sub> single crystal at ~20 and ~80 K from resistivity and thermoelectric power measurements (14–16). Surzuki *et al.* subsequently showed that the transition at ~80 K in Nb<sub>3</sub>Te<sub>4</sub> is associated with CDW formation (17–18). In 1987, Sekine *et al.* also observed two anomalies at 40 and 110 K in both the resistivity and the magnetic susceptibility data, where the 110 K phase transition is accompanied by the appearance of superlattice reflections which are characterized by commensurate wave vectors  $\mathbf{q} = \pm (\frac{1}{3}\mathbf{a}^* + \frac{1}{3}\mathbf{b}^*) + \frac{3}{7}\mathbf{c}^*$  (19).

Calculations of the electronic band structures of  $Nb_3Q_4$ by several groups indicate a quasi-one-dimensional character of these phases (9, 20, 21). The Fermi level of  $Nb_3S_4$ lies below a minimum of the density of states (DOS) curve. However, the position of the Fermi level can be altered by addition/removal of electrons in the  $Nb_3S_4$  structure.

The first ternary niobium chalcogenides  $A_x Nb_6 S_8$  (A = Li, Na, K, Ca) were prepared both (i) electrochemically and (ii) by solid state reaction by Schölhorn and Schramm (22). They found that the superconducting transition of Nb<sub>3</sub>S<sub>4</sub> increased:  $T_c = 7.3$  K for K<sub>0.4</sub>Nb<sub>6</sub>S<sub>8</sub>.

Huan and Greenblatt subsequently prepared many isotypic compounds by means of molten salt ion exchange or ternary element insertion methods:  $A_x Nb_6 S_8$  (A = Na, K, Rb, Tl, Cu, Ag, Zn, Pb, and Bi;  $0.67 \le x \le 1.0$ ),  $A_x Nb_6 Se_8$ (A = Tl, Sn, Pb, and Bi;  $0.4 \le x \le 1.0$ ), and  $A_x Nb_6 Te_8$ (A = Na, K, Rb, Cs, Tl, Cu, Ag, In, Zn, Cd, Sn, Pb, and Bi;  $0.67 \le x \le 2.0$ ) (10, 23). They also observed a

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Niobium	Sulfur
0 z = 1/4	O z = 1/4
• $z = 3/4$	$\square z = 3/4$

**FIG. 1.** Structure of  $A_x Nb_6 O_8$  (Q = S, Se, Te; A = alkali, alkaline earth, Tl, etc.). Shaded and unshaded octahedra are on level  $z = \frac{1}{4}$  and  $z = \frac{3}{4}$ , respectively.

considerable increase in  $T_c$  with the insertion of ternary elements into the empty tunnels of Nb<sub>3</sub>Q<sub>4</sub>: for example,  $T_c = 4.3$ , 5.8, and 6.3 K for A = Tl, Bi, and Pb in  $A_x$ Nb<sub>6</sub>Te<sub>8</sub> (24).

Recently, Ohtani *et al.* have prepared  $In_xNb_3Q_4$  ( $0 \le x$  1.0; Q = S, Se, Te) by the ternary element insertion method (11). All the polycrystalline samples show metallic behavior in the temperature range 1.5–273 K, with an anomalous jump of resistivity at 40–140 K for tellurides with x = 0.10-1.0. Electron diffraction measurements showed the 140 K transition to be associated with a CDW formation evidenced by the appearance of a superstructure with the commensurate wave vectors  $q = \pm (\frac{1}{3}a^* + \frac{1}{3}b^*) + \frac{4}{3}c^*$ .

Since the first report of ternary niobium chalcogenides  $A_x Nb_6 S_8$  (A = Li, Na, K, and Ca) by Schölhorn and Schramm, there have been many isotypic compounds prepared. However, alkaline earth ( $M^{2+}$ ) and rare earth ( $Ln^{3+}$ ) cations in  $A_x Nb_6 Te_8$  have not been investigated systematically (10, 23). As already predicted both from theoretical calculations and experimental results, it is feasible to insert both di- and trivalent ions into the empty channels of Nb<sub>3</sub>Te<sub>4</sub>-type compounds (9). In the present paper, we report some new ternary niobium chalcogenides of the type  $A_x Nb_6 Te_8$ , where A = Ca, Sr, Ba, La, Nd, and Tl, and their electrical and magnetic properties.

#### SYNTHESIS

#### Preparation of Tl<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub> by Solid State Method

The ternary  $Tl_xNb_6Te_8$  phases were prepared according to the procedure reported earlier (10, 23). First, TITe were prepared by heating pellets of a stoichiometric mixture of elemental Tl (Aldrich, 99.999%) and Te (Alfa, 99.99%) in a sealed quartz tube at 450°C for 4 h. Next, stoichiometric amounts of (TITe + 6 Nb + 7 Te) were mixed and pressed into pellets. The pellets were placed in a quartz tube, which was subsequently sealed under high vacuum,  $\sim 10^{-6}$  Torr, and heated at 1000°C for 10 days.

# Preparation of $A_xNb_6Te_8$ (A = Ca, Sr, Ba, La, and Nd) by Molten Salt Ion Exchange Reaction

The molten salt ion exchange reactions were carried out according to the equation

$$Tl_xNb_6Te_8 + yAI_3/AI_2 \rightarrow A_zNb_6Te_8 + xTII + excess AI_3/AI_2.$$

The powdered mixture of either Tl<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub> and *y* AI<sub>3</sub> (LaI<sub>3</sub>, 99.9%; NdI<sub>3</sub>, 99.99%; Alfa; y = 1.5-2) or Tl<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub> and 2 AI<sub>2</sub> (CaI<sub>2</sub>, 99.999%; SrI<sub>2</sub>, 99.998%; BaI<sub>2</sub>, 99.999%; Alfa) was mixed and pressed into a pellet under helium atmosphere and sealed in a quartz tube under helium atmosphere and sealed quartz tube was placed in a horizontal tube furnace with one end close to room temperature. Initially, the temperature of the sample in the hot zone was increased up to 450°C and held for 3–4 days. The set temperature 450°C is ~10°C higher than the melting point of TII. The tube was then heated at the final reaction temperatures, 804, 535, 760, 790, and 795°C for Ca, Sr, Ba, La, and Nd, respectively, which are approximately ~10–20°C higher than the melting point of corresponding iodides, AI<sub>3</sub> or AI<sub>2</sub>.

#### CHARACTERIZATION

X-ray powder diffraction patterns of all the samples were recorded with a SCINTAG PAD V diffractometer with CuK $\alpha$  radiation in the 2 $\theta$  range 5°–65°. Si or W powders were used as internal standards and cell parameters were obtained from least-square refinement of the observed d spacings.

The dc electrical resistivities were measured on brickshaped polycrystalline pellet with a DE 202 cryostat (APD Cryogenics) by a four-probe method between 25 and 300 K. Ohmic contacts were made either by attaching molten indium ultrasonically to the four spots of conducting silver paste on the sides of the specimens or by anchoring thin copper wires (diameter  $\sim 0.04$  mm) on the sides of the specimens with a conductive Ag paint.

The magnetic susceptibility of the samples was measured in a SQUID magnetometer (Quantum Design) in the temperature range  $2 \sim 300$  K at applied magnetic fields  $25 \sim$ 1000 G. Brick shaped specimens, similar to those used for the resistivity measurements, were employed in the magnetic property investigations.

Eu, una r(a) i nasos								
Compound	a (Å)	<i>c</i> (Å)	c/a	$V(\text{\AA}^3)$				
Nb <sub>3</sub> Te <sub>4</sub> <sup><i>a</i></sup>	10.671(1)	3.6468(1)	0.3418	359.6(1)				
Na <sub>0.86</sub> Nb <sub>6</sub> Te <sub>8</sub> <sup>b</sup>	10.653(1)	3.642(2)	0.3419	357.8(1)				
Zn <sub>1.0</sub> Nb <sub>6</sub> Te <sub>8</sub> <sup>b</sup>	10.651(1)	3.639(1)	0.3437	357.5(1)				
Bi <sub>0.67</sub> Nb <sub>6</sub> Te <sub>8</sub> <sup>b</sup>	10.648(1)	3.636(1)	0.3415	357.0(1)				
Tl <sub>0.91</sub> Nb <sub>6</sub> Te <sub>8</sub>	10.673(1)	3.652(1)	0.3422	360.3(1)				
La <sub>0.32</sub> Nb <sub>6</sub> Te <sub>8</sub>	10.680(1)	3.637(1)	0.3405	359.3(1)				
Nd <sub>0.24</sub> Nb <sub>6</sub> Te <sub>8</sub>	10.672(1)	3.647(2)	0.3417	359.7(2)				
$Ca_xNb_6Te_8$	10.670(1)	3.647(1)	0.3418	359.6(1)				
Sr <sub>0.22</sub> Nb <sub>6</sub> Te <sub>8</sub>	10.699(1)	3.630(1)	0.3393	359.9(1)				
$Ba_{0.24}Nb_6Te_8$	10.668(2)	3.649(2)	0.3421	359.6(1)				

TABLE 1 Unit Cell Parameters of  $A_x Nb_6 Te_8$  (A = Na, Tl,La, and Nd) Phases

*Note.* Space group =  $P6_3/m$ . Internal standard = W.

<sup>*a*</sup> Calculated from a and c in Ref. (7).

<sup>b</sup> Calculated from a and c in Ref. (24).

A microprocessor-controlled dc plasma atomic emission spectrometer (Spectrametrics, Inc., SMI III) was used for the composition analysis of the samples.

## **RESULTS AND DISCUSSION**

Table 1 lists the unit cell parameters of pure  $Tl_xNb_6Te_8$ and of the new  $A_x Nb_6 Te_8$  (A = Ca, Sr, Ba, La, and Nd) phases. Polycrystalline  $Tl_xNb_6Te_8$  and  $A_xNb_6Te_8$  were prepared by solid state reactions and by molten salt ion exchange reactions, respectively. All the ternary niobium chalcogenides were synthesized in the pure form. However, powder X-ray diffraction patterns of new La<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub> and  $Nd_xNb_6Te_8$  sometimes show additional weak peaks at  $\sim 8.8^{\circ}$ , 36.3°, and 50.2°, probably due to either lowering of symmetry or an unknown impurity phase. The unit cell parameters of all new tellurides (A = Ca, Sr, Ba, La, andNd) were similar to those of the binary Nb<sub>3</sub>Te<sub>4</sub> and the starting compound Tl<sub>0.91</sub>Nb<sub>6</sub>Te<sub>8</sub>, suggesting that the unit cell parameters were not affected by the size of the A ions. However, this is not surprising because the Nb<sub>3</sub>Te<sub>4</sub> structure has a rigid skeleton due to the stable Nb-Nb and Nb-Te bonds. Similar results were observed in many other ternary chalcogenides (10, 11, 23). The observed and calculated d spacings of  $A_x Nb_6 Te_8$  are compared in Table 2.

Plasma atomic emission (PAE) measurements showed that the x values are 0.91, 0.32, 0.24, 0.24, and 0.22, for Tl, La, Nd, Ba, and Sr, respectively within an experimental error ~10%. We could not determine the Ca content in  $Ca_xNb_6Te_8$  samples due to a precipitation in the solution. The PAE also confirmed that no Tl remained in the powdered samples after the ion exchange reaction of  $Tl_xNb_6Te_8$ by La, Nd, Ba, Sr, and Ca, indicating no mixed occupancy at the A site in  $A_xNb_6Te_8$  compositions.

The temperature dependence of the electrical resistivi-

ties of  $A_r Nb_6 Te_8$  in the range 25–300 K are shown in Figs. 2 and 3. All the samples show metallic behavior over the entire temperature range of measurement with anomalous increases of resistivity at low temperature. Tl<sub>0.91</sub>Nb<sub>6</sub>Te<sub>8</sub> shows a broad transition at 160-220 K (Fig. 2a). This transition was also observed in differential scanning calorimetric analysis, where a small endothermic peak was observed at 163 K, suggesting a possible minor structural distortion. The resistivity anomaly at 160-220 K might be due to a CDW-type instability. Magnetic susceptibility measurements of Tl<sub>0.91</sub>Nb<sub>6</sub>Te<sub>8</sub> indicate typical Pauli paramagnetism in the temperature range 2-300 K and show no evidence of superconductivity down to 2 K. This is in contrast to Huan et al.'s observation of a superconducting transition at 4.3 K for TlNb<sub>6</sub>Te<sub>8</sub> by electrical resistivity measurement (24). Presently, it is not clear why the  $Tl_{0.91}Nb_6Te_8$  samples studied here are not superconducting. It is possible that the superconductivity is suppressed by a CDW formation in our Tl<sub>0.91</sub>Nb<sub>6</sub>Te<sub>8</sub> samples, since the density of states near the Fermi level decrease with a CDW formation. The absence of superconductivity in the Tl<sub>0.91</sub>Nb<sub>6</sub>Te<sub>8</sub> samples might be explained in terms of the competition between superconductivity and CDW instabilities.

La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub> (Fig. 2b) and Nd<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> (Fig. 2c) show metallic behavior in the temperature range 25-300 K, with an anomalous transition at 46 and 43 K, respectively. A very small hysteresis was observed upon heating and cooling for both compounds. These transitions in the A = Laand Nd phases are most likely due to CDW instabilities, similar to that observed in Nb<sub>3</sub>Te<sub>4</sub> single crystals by Sekine et al. (19). They observed an anomaly at  $\sim 40$  K both in the resistivity and magnetic susceptibility, although superlattice lines in the electron diffraction patterns were not observed below 40 K in Nb<sub>3</sub>Te<sub>4</sub>. Both La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub> and Nd<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> show an unusual nonlinear behavior of electrical resistivity with a positive curvature in the range 25-300 K. The resistivity near room temperature decreases slowly, while at low temperature it decreases rapidly with decreasing temperature. This broad deviation of resistivity from linearity could be due to disorder of the A cations in the tunnels.

 $Ca_xNb_6Te_8$ ,  $Sr_{0.22}Nb_6Te_8$ , and  $Ba_{0.24}Nb_6Te_8$  also show metallic behavior in the temperature range 25–300 K (Fig. 3). These compounds do not show any resistivity anomaly down to 30 K except a broad positive curvature, which is similar to that observed in  $La_{0.32}Nb_6Te_8$ .

The magnetic susceptibility of La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub> (Fig. 4) revealed a sharp superconducting transition ( $T_c^{\text{onset}} = 4$  K) and Ca<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub>, Sr<sub>0.22</sub>Nb<sub>6</sub>Te<sub>8</sub>, and Ba<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> showed a broad superconducting transition ( $T_c^{\text{onset}} = 6$  K) with typical Pauli-paramagnetic behavior above these transition temperatures (Figs. 4 and 5). The  $T_c$ 's observed at 4–6 K for  $A_x$ Nb<sub>6</sub>Te<sub>8</sub> (A =Ca, Sr, Ba, and La) are significantly higher than that of Nb<sub>3</sub>Te<sub>4</sub> at 1.8 K. However, Nd<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub>

TABLE 2Powder X-Ray Diffraction Data of  $A_x$ Nb<sub>6</sub>Te<sub>8</sub> Phases (A = Tl, La, Nd, Ca, Sr, and Ba)

	Т	$Tl_{0.91}Nb_6Te_8$		L	$La_{0.32}Nb_6Te_8$			$Nd_{0.24}Nb_6Te_8$		
hkl	$d_{ m obs}({ m \AA})$	$d_{\mathrm{calc}}(\mathrm{\AA})$	I/I <sub>o</sub>	$\overline{d_{ m obs}}({ m \AA})$	$d_{\mathrm{calc}}(\mathrm{\AA})$	I/I <sub>o</sub>	$d_{ m obs}({ m \AA})$	$d_{\mathrm{calc}}(\mathrm{\AA})$	I/I <sub>o</sub>	
100	9.26	9.24	42	9.27	9.26	89	9.24	9.24	78	
110	5.338	5.337	7	5.342	5.345	10	5.338	5.336	11	
210	3.494	3.494	3	3.499	3.499	4	3.490	3.493	7	
300	3.081	3.081	72	3.087	3.086	72	3.081	3.081	61	
111	3.014	3.014	7	3.009	3.007	6	3.014	3.011	2	
201		2.865		2.859	2.860	20	2.863	2.863	6	
310	2.564	2.564	99	2.568	2.568	99	2.565	2.563	99	
211	2.524	2.525	23	2.522	2.522	42	2.523	2.523	20	
301		2.355		2.353	2.353	5		2.353		
400	2.310	2.311	2		2.314		2.311	2.311	5	
221	2.154	2.155	11	2.154	2.154	19	2.154	2.153	9	
320	2.120	1.121	10	2.123	2.124	10	2.120	2.120	12	
311	2.099	2.098	7	2.098	2.098	12	2.096	2.097	4	
410	2.017	2.017	20	2.020	2.020	19	2.017	2.017	20	
500	1.849	1.849	13	1.851	1.852	16	1.848	1.849	21	
321		1.834		1.834	1.834	16	1.835	1.833	9	
420	1.746	1.747	11	1.749	1.750	15	1.746	1.747	13	
510	1.660	1.660	5		1.663		1.659	1.660	6	
501		1.649		1.650	1.650	9	1.648	1.649	5	
331		1.599		1.600	1.600	5	1.598	1.599	4	
430	1.520	1.520	2		1.522			1.520		
312		1.487			1.484		1.485	1.486	4	
520		1.480		1.482	1.482	12	1.479	1.480	9	
	(	$Ca_xNb_6Te_8$		$Sr_{0.22}Nb_6Te_8$			$Ba_{0.24}Nb_6Te_8$			
hkl	$\overline{d_{ m obs}}({ m \AA})$	$d_{ m calc}$ (Å)	I/I <sub>o</sub>	$\overline{d_{ m obs}}({ m \AA})$	$d_{ m calc}$ (Å)	I/I <sub>o</sub>	$d_{ m obs}({ m \AA})$	$d_{ m calc} ({ m \AA})$	I/I <sub>o</sub>	
100	9.245	9.241	73	9.274	9.266	58	9.256	9.239	58	
110	5.335	5.335	12	5.341	5.350	9	5.341	5.334	11	
210		3.493			3.502			3.492		
300	3.081	3.080	70	3.089	3.089	62	3.081	3.080	43	
111	3.010	3.011	6	3.002	3.004	15	3.013	3.012	22	
201	3.862	3.862	15	2.858	2.858	48	2.864	2.863	54	
310	2.563	2.563	99	2.570	2.570	99	2.564	2.562	99	
211	2.523	2.522	22	2.521	2.520	89	2.523	2.523	89	
301		2.353		2.351	2.352	13		2.354		
400	2.310	2.310	8		2.316			2.310		
221	2.154	2.153	12	2.154	2.153	41	2.154	2.153	40	
221	2.134				2.126			2.120		
320	2.134	2.120			2.120				23	
	2.134	2.120 2.097	8	2.098	2.098	27	2.097	2.097	25	
320			8 32	2.098 2.021		27 26	2.097 2.014	2.097 2.016	25 25	
320 311	2.096	2.097			2.098					
320 311 410	2.096 2.016	2.097 2.016	32	2.021	2.098 2.022	26	2.014	2.016	25	
320 311 410 500	2.096 2.016 1.848	2.097 2.016 1.848	32 28	2.021 1.853	2.098 2.022 1.853	26 18	2.014 1.848	2.016 1.848	25 17	
320 311 410 500 321	2.096 2.016 1.848 1.833	2.097 2.016 1.848 1.833	32 28 12	2.021 1.853 1.835	2.098 2.022 1.853 1.834	26 18 34	2.014 1.848 1.833	2.016 1.848 1.833	25 17 36	
320 311 410 500 321 420	2.096 2.016 1.848 1.833	2.097 2.016 1.848 1.833 1.746	32 28 12	2.021 1.853 1.835	2.098 2.022 1.853 1.834 1.751	26 18 34	2.014 1.848 1.833	2.016 1.848 1.833 1.746	25 17 36	
320 311 410 500 321 420 510	2.096 2.016 1.848 1.833 1.746	2.097 2.016 1.848 1.833 1.746 1.660	32 28 12 19	2.021 1.853 1.835 1.751	2.098 2.022 1.853 1.834 1.751 1.664	26 18 34 19	2.014 1.848 1.833 1.744	2.016 1.848 1.833 1.746 1.659	25 17 36 14	
320 311 410 500 321 420 510 501	2.096 2.016 1.848 1.833 1.746 1.649	2.097 2.016 1.848 1.833 1.746 1.660 1.649	32 28 12 19 9	2.021 1.853 1.835 1.751 1.650	2.098 2.022 1.853 1.834 1.751 1.664 1.651	26 18 34 19 22	2.014 1.848 1.833 1.744 1.648	2.016 1.848 1.833 1.746 1.659 1.649	25 17 36 14 24	
320 311 410 500 321 420 510 501 331	2.096 2.016 1.848 1.833 1.746 1.649	2.097 2.016 1.848 1.833 1.746 1.660 1.649 1.598	32 28 12 19 9	2.021 1.853 1.835 1.751 1.650	2.098 2.022 1.853 1.834 1.751 1.664 1.651 1.601	26 18 34 19 22	2.014 1.848 1.833 1.744 1.648	2.016 1.848 1.833 1.746 1.659 1.649 1.598	25 17 36 14 24	

*Note.* Space group =  $P6_3/m$ . Internal standard = W powder.

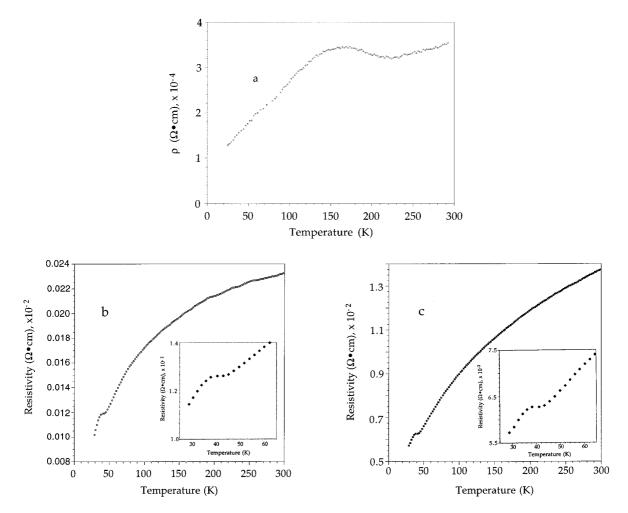
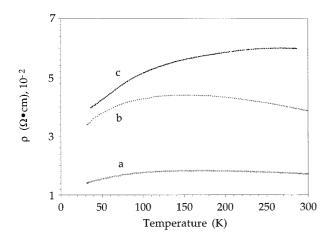


FIG. 2. Temperature dependence of electrical resistivity of (a)  $Tl_{0.91}Nb_6Te_8$ , (b)  $La_{0.32}Nb_6Te_8$ , and (c)  $Nd_{0.24}Nb_6Te_8$ . There is a broad resistivity anomaly at 160–220 K for  $Tl_{0.91}Nb_6Te_8$  and a resistivity anomaly at 46 and 43 K for  $La_{0.32}Nb_6Te_8$  and  $Nd_{0.24}Nb_6Te_8$ , respectively.



**FIG. 3.** Temperature dependence of electrical resistivity of (a)  $Ca_xNb_6Te_8$ , (b)  $Sr_{0.22}Nb_6Te_8$ , and (c)  $Ba_{0.24}Nb_6Te_8$ .

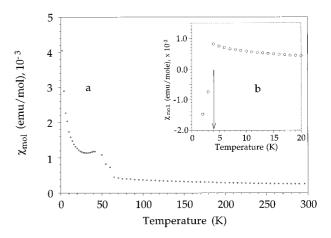
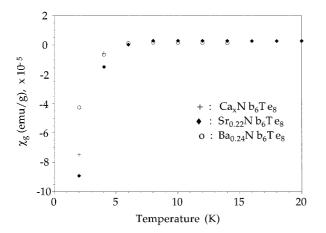
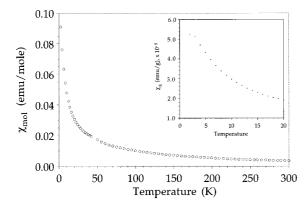


FIG. 4. Magnetic susceptibility of  $La_{0.32}Nb_6Te_8$  in the temperature range 2–300 K. The inset shows a sharp superconducting transition at 4 K ( $T_c^{\text{onset}}$ ).



**FIG. 5.** Magnetic susceptibility measurements of  $Ca_xNb_6Te_8$ ,  $Sr_{0.22}Nb_6Te_8$ , and  $Ba_{0.24}Nb_6Te_8$  in the temperature range 2–300 K. All show a superconducting transition at ~6 K ( $T_c^{noset}$ ).

did not show any evidence of superconductivity down to 2 K. The absence of superconductivity in Nd<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> may be attributed to the highly paramagnetic Nd<sup>3+</sup> ions in Nd<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> (Fig. 6). The inset shows a slower increase of susceptibility at ~3 K, probably due to either the onset of a short-range antiferromagnetic correlation between the Nd<sup>3+</sup> moments, or the formation of a filamentary superconducting state. Detailed ac susceptibility and electrical resistivity measurements below 4 K would be needed to further clarify this anomaly. Provided that the temperature-independent magnetic susceptibility ( $\chi_0$ ) is negligible as seen in La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub>, the Curie (C = 0.87 mole/emu · K) and Weiss ( $\theta = 28$  K) constants were obtained by analyzing the susceptibility data in the temperature range 200–300 K, following a modified Curie–Weiss equation,



 $\chi = \chi_{\rm p} + C/(T - \theta),$ 

**FIG. 6.** Magnetic susceptibility of  $Nd_{0.24}Nb_6Te_8$  in the temperature range 2–300 K. The inset shows a slower increase of susceptibility at ~3 K, probably due to either a short-range antiferromagnetic coupling or a formation of filamentary superconducting state.

where  $\chi_{\rm p}$  is the Pauli susceptibility due to conduction electrons. Pauli susceptibility was estimated based on the temperature-independent susceptibility (100–300 K) observed for La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub> (Fig. 4). The effective magnetic moment ( $\mu_{\rm B} = 2.63$ ) would correspond to x = 0.51 in Nd<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub>, assuming spin-only contribution of 4f<sup>3</sup> electrons, which is somewhat higher than the value estimated from PAE analysis (x = 0.24). However, strong covalent interactions between Nd and Te might be responsible for the observed magnetic data.

## CONCLUSION

The ternary niobium chalcogenides,  $A_x Nb_6 Te_8$  (A = Ca, Sr, Ba, La, and Nd) were prepared by molten salt ion exchange reactions of  $Tl_{0.91}Nb_6Te_8$  and the corresponding iodides (CaI<sub>2</sub>, SrI<sub>2</sub>, BaI<sub>2</sub>, LaI<sub>3</sub>, and NdI<sub>3</sub>) at low temperatures, 535–795°C. La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub>, Ca<sub>x</sub>Nb<sub>6</sub>Te<sub>8</sub>, Sr<sub>0.22</sub>Nb<sub>6</sub>Te<sub>8</sub>, and Ba<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> show a superconducting transition at 4–6 K, while  $Tl_{0.91}Nb_6Te_8$  and  $Nd_{0.24}Nb_6Te_8$  do not show any evidence of diamagnetism down to 2 K.  $Tl_{0.91}Nb_6Te_8$  shows a broad resistivity anomaly at 160–220 K, which is also evidenced by DSC analysis. Both La<sub>0.32</sub>Nb<sub>6</sub>Te<sub>8</sub> and Nd<sub>0.24</sub>Nb<sub>6</sub>Te<sub>8</sub> exhibit an anomaly in the temperature dependence of resistivity at ~46 and ~43 K, respectively, which suggest a CDW instability similar to that observed in Nb<sub>3</sub>Te<sub>4</sub> single crystals at ~40 K.

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